Summary of previous lecture

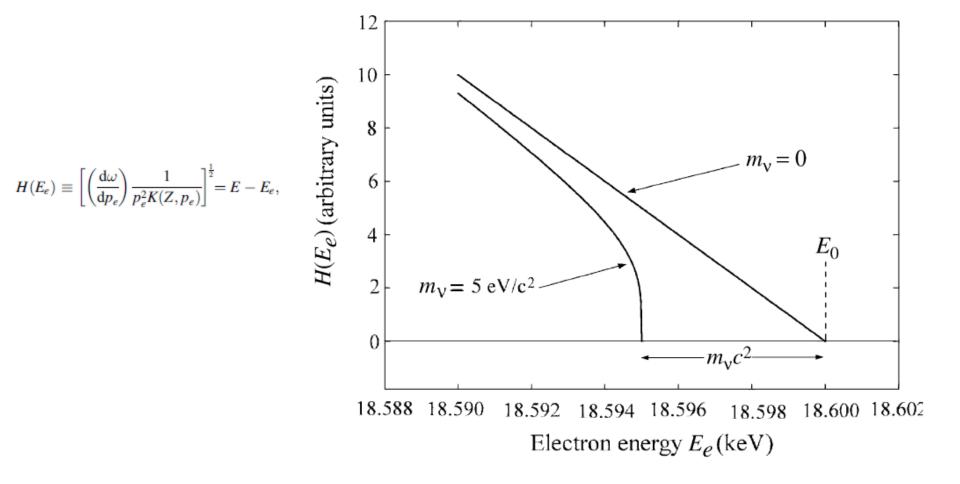
• We discussed α -decay theory, β -decay theory (Fermi-theory) and γ -ray theory

 We scetched the derivation of the electron spectrum for β-decay from Fermi-theory (main assumption: point-like interactions, short range Yukawa coupling, no Fermi screening), which is determined by the phase space factor.

Summary of previous lecture

- Geiger-Nutall relation for α-decay
 relates half lives to Q-value
- Fermi-Kurie relation for β-decay
 - deviation can point to "new" physics, e.g. neutrino mass
- Selection rules for γ decays
 - According to angular momentum and Parity conservation (n.b. internal conversion, internal pair production)

β decay and the neutrino: the Kurie plot



Lecture 5: Applications

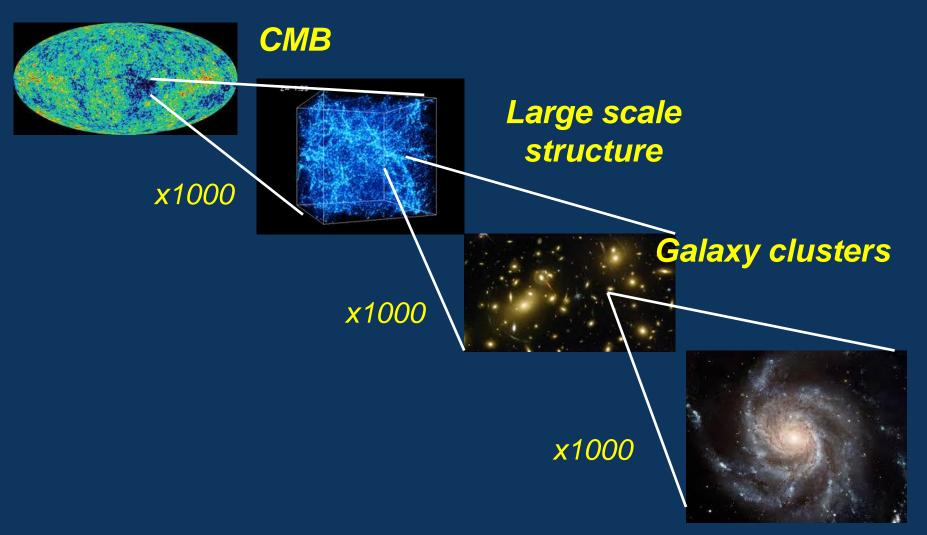
Jan Conrad

Overview

 Dark Matter Searches with rare event experiments ← impact of nuclear physics

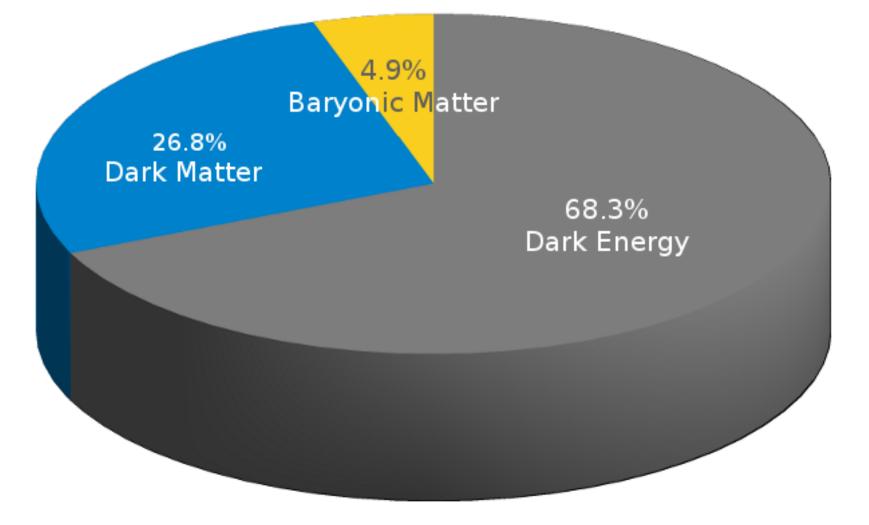
• Atomic bombs and nuclear reactors

EFFECT PRESENT ON ALL SCALES



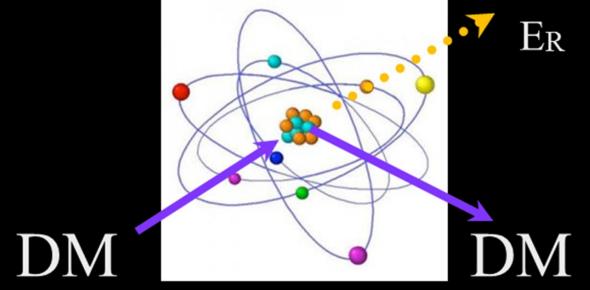
New Type(s) of Particle(s): → compelling, simple, (so far only) *valid* explanation.

The Universe as a pie chart



THE GENERAL IDEA

Elastic scattering of Dark matter on <u>atomic nucle</u>i

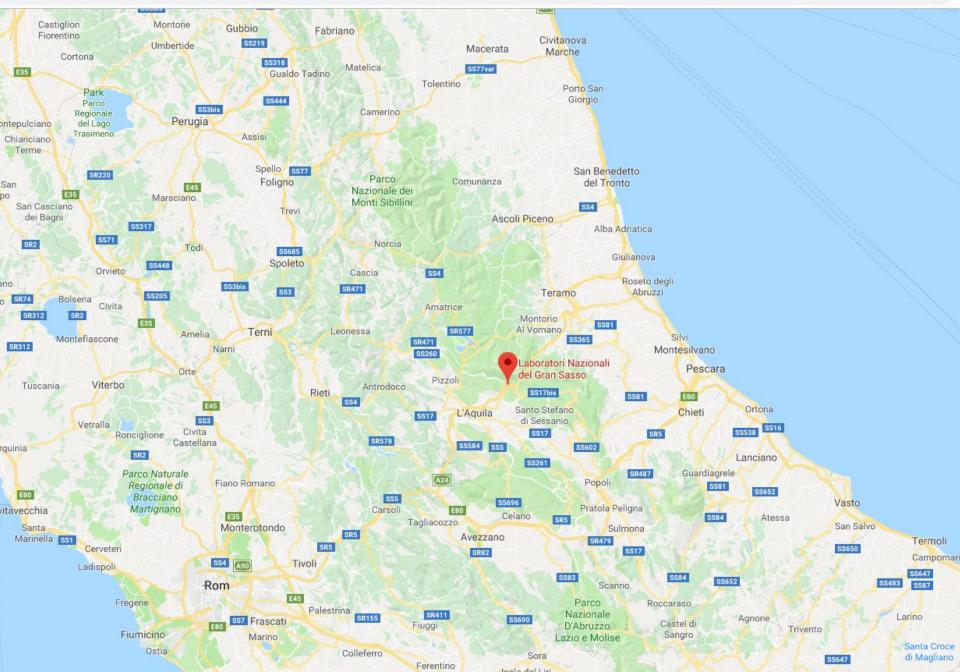


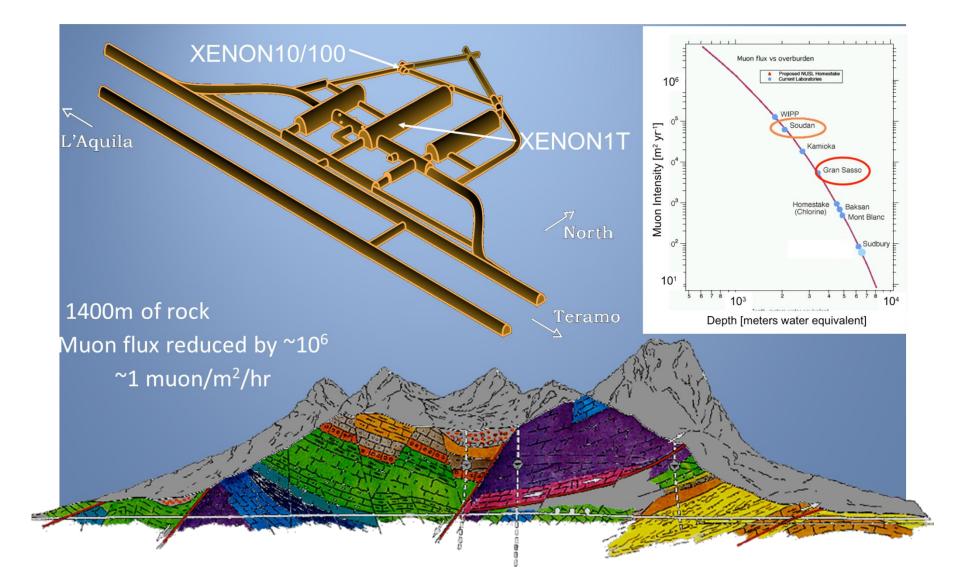
$$E_R = \frac{q^2}{m_n} < 100 \text{ keV}$$



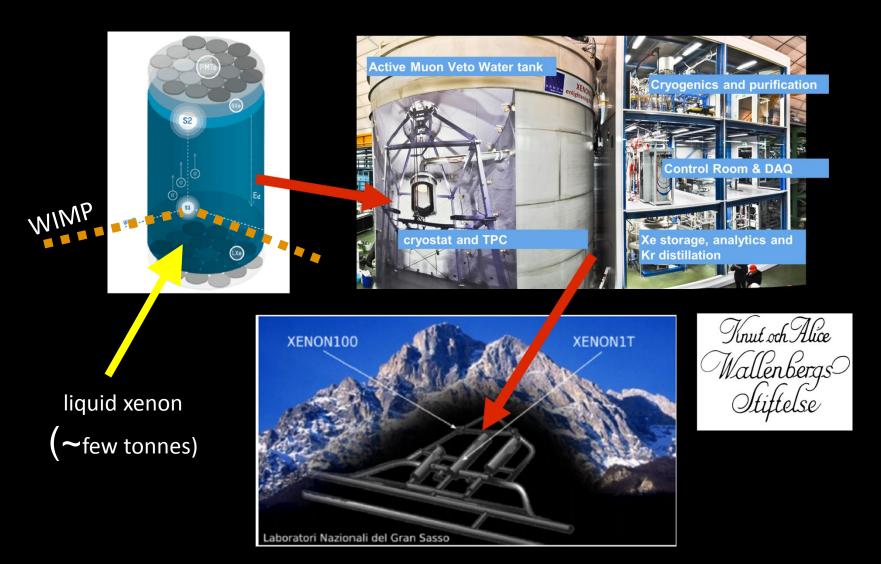
☆

lel+Gran+Sasso/@42.4202734,12.9656644,9z/data=l4m5!3m4l1s0x0:0xa264f2d2892359fcl8m2!3d42.4198187!4d13.5172445





DIRECT DETECTION: XENON1T@LNGS



Need to identify one interaction in 10 trillion!

The world's most sensitive dark matter detector



Natural radioactivity

Radioactive substances in all materials and in the environment

 Natural radioactivity is on the order of 10⁻³ Bq m³ of air → 30000 decays/yr/m³

We expect about 1/yr/m³ from dark matter detection.

What to do?

→we'll attempt to use only the cleanest materials
 (i.e. we'll have to measure the nuclear decays
 happening in the materials before we use them)

→we'll try to get rid of the remaining nuclear decays by understanding how they look in the detector.

Backgrounds: nuclear physics

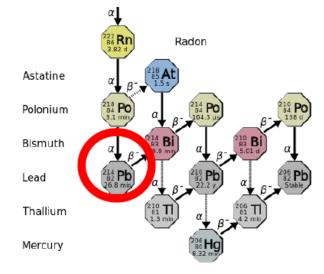
- Radon sticks on surfaces and emanates into the detector material.
- Rn-222: 10 µBq/kg,

1 decay /28 hours

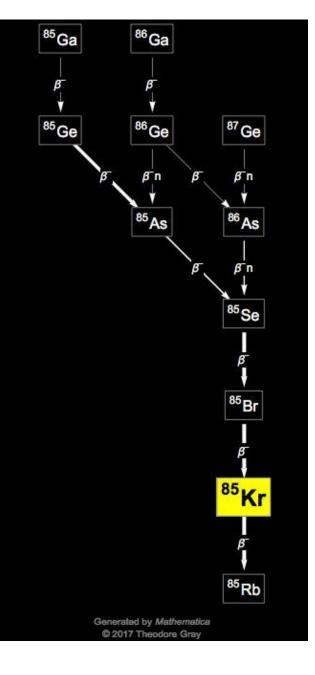
- Kr-85: sub-ppt Kr/Xe,
- Materials background: selected using HPGe gamma screening

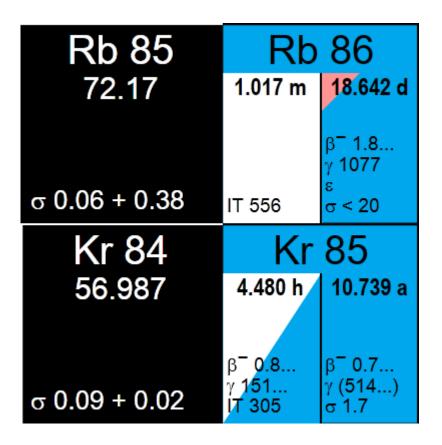
_____>

Radon and Krypton reduced using online cryogenic distillation.



Taken from a presentation of Bart Pelssers



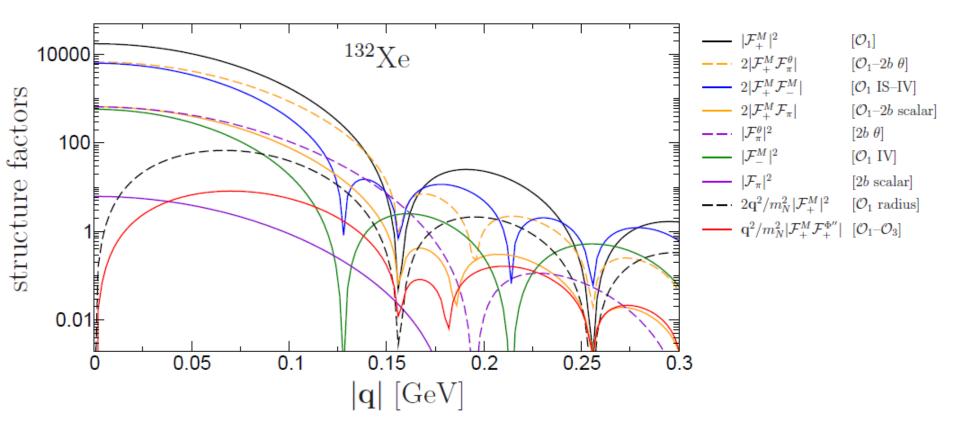


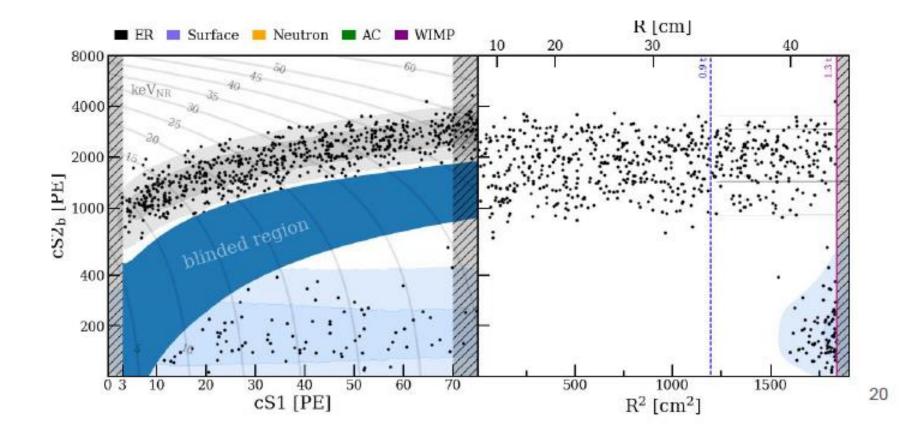
Nuclear isomer, IT Isomeric transition

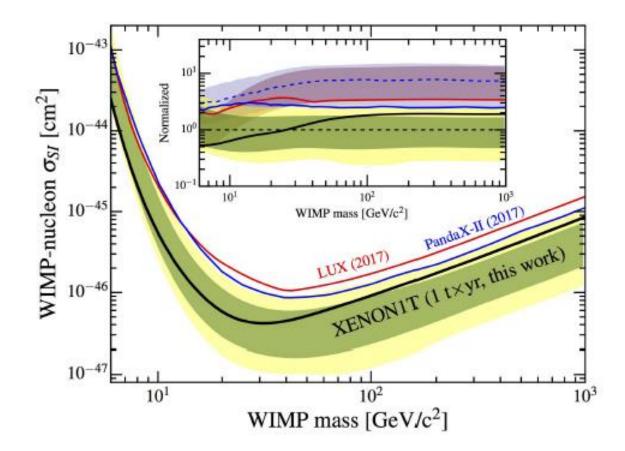
More nuclear physics?

 To learn about their properties of the dark matter particles, we'll have to understand how they would interact with the nucleus.

WIMP-nucleus interactions







Neutron induced fission

 Spontanous fission is inhibited by Coulumb barrier and does become the dominant decay mode only for very heavy nuclei (A>250)

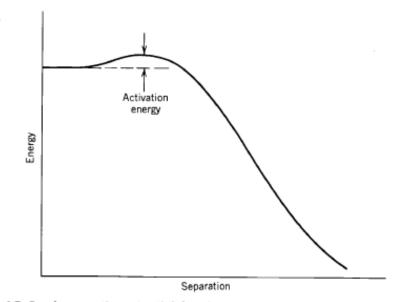
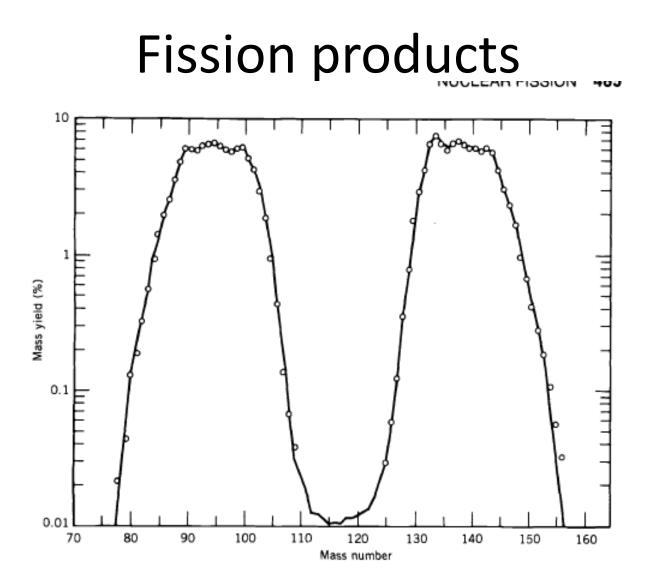


Figure 13.2 A smooth potential barrier opposing the spontaneous fission of ²³⁸U. To surmount the fission barrier, we must supply an amount of energy equal to the activation energy.

Induced fission

 A zero-energy neutron can form a compound nucleus with excitation energy above the Coulumb barrier → fission possible

 235 U+n \rightarrow 236 U \rightarrow 141 Ba+ 92 Kr+3n



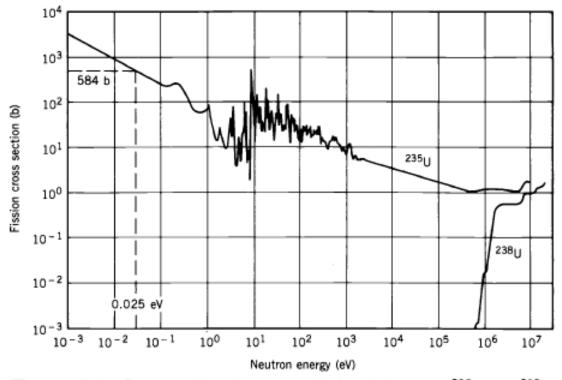
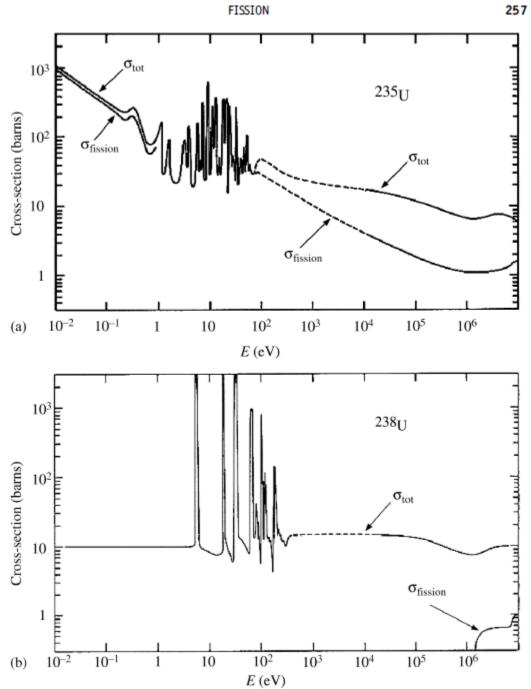
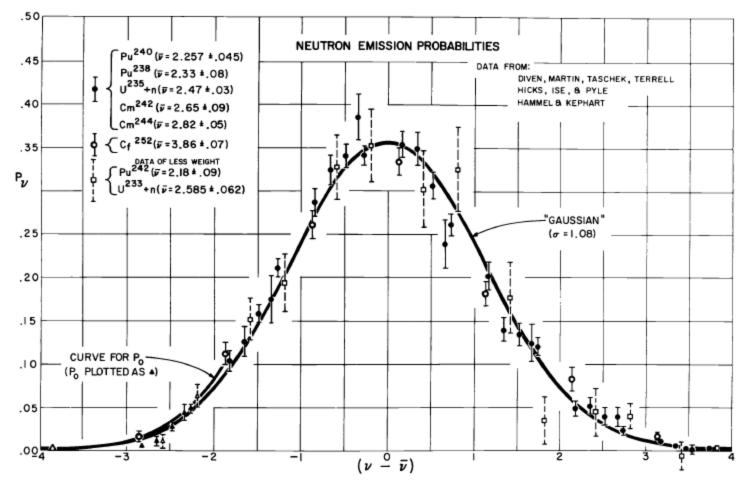


Figure 13.9 Cross sections for neutron-induced fission of ²³⁵U and ²³⁸U.





Neutron emission probabilities



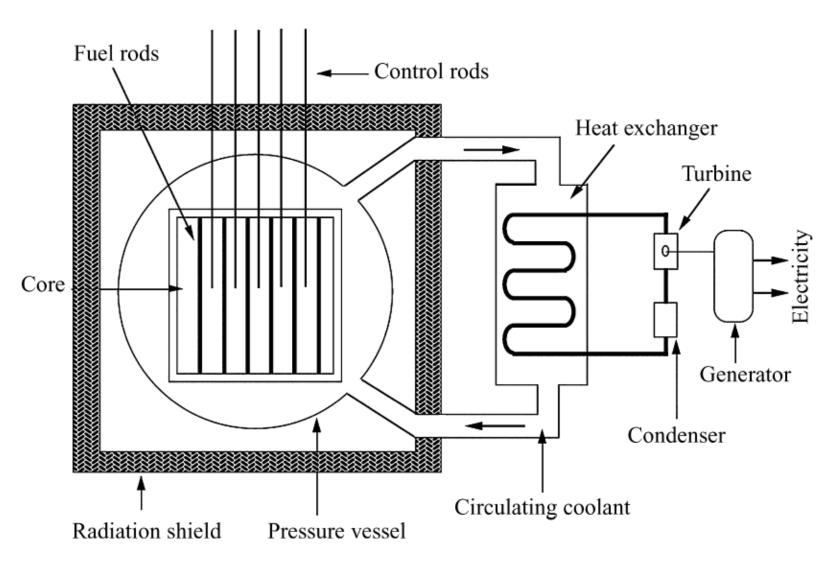
Fission chains

• We saw the distribution of produced neutrons. It makes sense to define the neutron reproduction factor:

 $k \equiv \frac{\text{number of neutrons produced in the } (n+1) \text{ th stage of fission}}{\text{number of neutrons produced in the } n \text{th stage of fission}}$

- k>1: supercritical, k=1 critical, k<1 sub-critical
- In natural Uranium, on average 2.5 neutrons are produced in each fission but most of them are fast → moderation

Nuclear reactor



Atomic bombs

• Atomic bombs are supercritical assemblies of fissile material.

 Trigger: sponteanous fission: e.g. U-235 will have about 15 spontaneous fission decays/second in its critical mass (see below).

$$l = \frac{1}{n6}$$

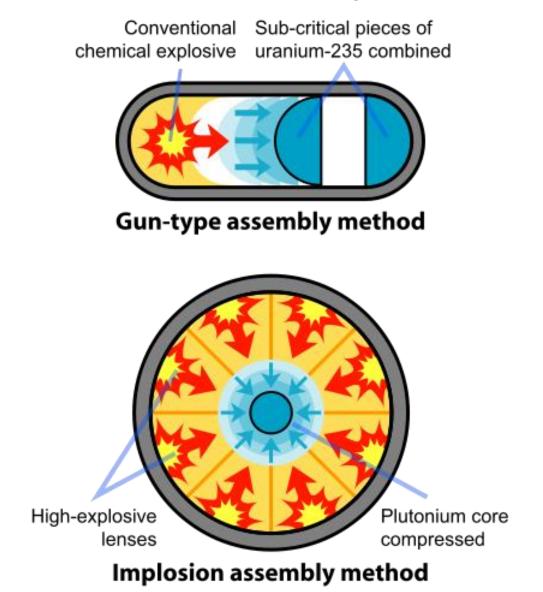
$$m = \frac{P_{u}}{A_{u}} \cdot N_{4} = \frac{\frac{16g}{a^{2}}}{3.35g} \frac{1}{mol} \cdot \frac{6 \cdot 10^{23}}{10^{22}}$$

$$\approx 5 \cdot 10^{22} / cm^{3}$$

$$\int L = \frac{1}{(2 \cdot 10^{-24} \text{ m}^2)(5 \times 10^{22} \text{ cm}^{-3})} \sim 10 \text{ cm}$$

Mass of ²³⁵U ophere with
$$r = 10 \text{ m}$$
:
 $\frac{4}{3}\pi \left(10 \text{ m}\right)^2 \left(\frac{16 \text{ g}}{\text{ m}^3}\right) \approx 60 \log D$

Assembly





https://soundcloud.com/atomicheritage/jrobert-oppenheimer

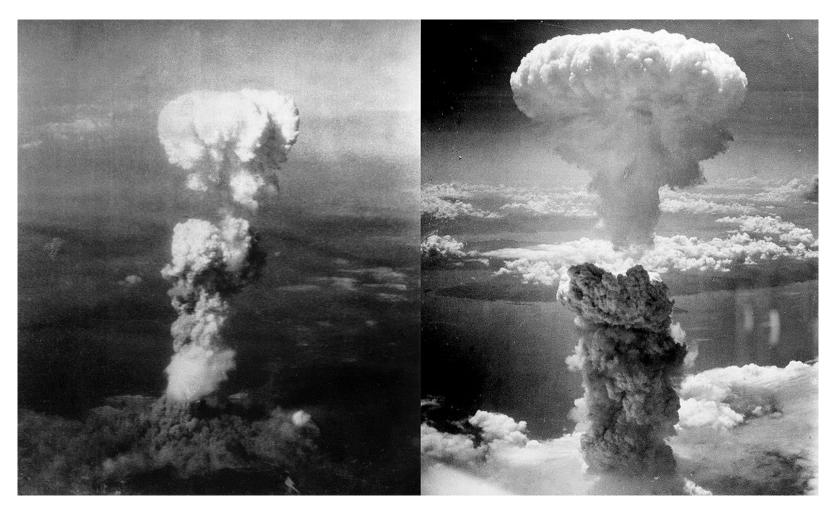
https://www.manhattanprojectvoices.org/

Nagasaki/Hiroshima



Little Boy and Fat Man

Nagasaki/Hiroshima



Hiroshima: gun-type, U235 bomb (16 kt TNT)

Nagasaki: implosion type, plutonium bomb (21-25 kt TNT)

Summary of today's lecture

• We discussed two applications of nuclear physics.