Lecture 10 – Lattice vibrations II

Reading

Ashcroft & Mermin, Ch. 23

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Central concepts

Normal modes

A general motion of N ions is represented as a superposition (linear combination) of 3N normal modes of vibration, each with its own characteristic frequency.

The allowed energies of an oscillator with frequency ν are given by

$$\left(n+\frac{1}{2}\right)h\nu, \qquad n=0,1,2,\dots \qquad (h\nu=\hbar\omega)$$

The 3N normal modes correspond to 3N oscillators, each with the above possible energies.

If the crystal has a primitive cell of p ions, there are 3p branches, each with $N_{cell} = N/p$ normal modes.

• Phonons

Instead of talking about the nth energy state of a certain normal mode of branch s and wave vector \mathbf{K} , one says that there is n phonons of type s with wave vector \mathbf{K} present in the crystal.

The number of phonons in each mode at thermal equilibrium is given by the Bose-Einstein distribution

$$n_s(\mathbf{K}) = \frac{1}{e^{\hbar \omega_s(\mathbf{K})/k_B T} - 1}$$

• Zero-point vibrations

Summing the energy of all oscillators gives a contribution

$$u_0 = \frac{1}{V} \sum_{\mathbf{K}s} \frac{1}{2} \hbar \omega_s(\mathbf{K})$$

to the energy density. This vibration energy of $\hbar\omega_s(\mathbf{K})/2$ per mode is present even at T=0.

• High-temperature specific heat

The temperature dependent part of the energy density of a harmonic crystal is given by summing the number of phonons $n_s(\mathbf{K})$ times their energy $\hbar\omega_s(\mathbf{K})$,

$$u = \frac{1}{V} \sum_{\mathbf{K}_s} \frac{\hbar \omega_s(\mathbf{K})}{e^{\hbar \omega_s(\mathbf{K})/k_B T} - 1}$$

At high temperatures $e^x \approx 1 + x$, so that $n_s = k_B T / \hbar \omega$. Then the expression becomes a sum over $k_B T$, so that $u \approx 3nk_B T$ (n = N/V). This gives the classical Dulong-Petit specific heat

$$c_v = \frac{\partial u}{\partial T} = 3nk_B$$

Anharmonic effects are the main reason for deviations at high T from this value.

• Debye model

The Debye model makes two assumptions

- $\star \omega = v_s |K|$, where v_s is the sound velocity
- ★ Introduce $\omega_D = v_s K_D$ such that a sphere of radius K_D contains N_{ion} allowed wave vectors (corresponding to 3N states in 3D with three polarization directions).

Since the (reciprocal) volume per **K**-point is $(2\pi)^3/V$, the second assumption gives

$$K_D = (6\pi^2 n)^{1/3}$$

where n = N/V.

• Debye temperature

The Debye temperature Θ_D is defined from

$$\hbar\omega_D=k_B\Theta_D$$

• Low-temperature specific heat

Using the Debye model, letting

$$\sum_{\mathbf{K}} \to \frac{V}{(2\pi)^3} \int d\mathbf{K}$$

and letting $d\mathbf{K} = 4\pi K^2 dK$, the heat capacity at low temperature can be written as

$$c_v \approx 234 \left(\frac{T}{\Theta_D}\right)^3 n k_B$$

• Einstein model

The Einstein model makes the assumption that all modes have the same frequency, i.e.,

$$\omega(\mathbf{K}) = \omega_E$$

This assumption is thus best suited for the optical branches, where ω does not go to zero at long wavelengths (small K). Similarly to Θ_D an Einstein temperature Θ_E can be defined, $\hbar\omega_E = k_B\Theta_E$.

• Density of normal modes

It is useful to introduce the density of normal modes $D(\omega)$ corresponding to the number of modes between ω and $\omega + d\omega$. As earlier, we define $g(\omega) = D(\omega)/V$.

With the Debye model, this gives

$$g_D(\omega)d\omega = 3\frac{1}{(2\pi)^3}4\pi K^2 dK$$

i.e. (Eq. 23.36)

$$g_D(\omega) = \frac{3}{2\pi^2} \frac{\omega^2}{v_s^3}$$

for frequencies up to ω_D .

For the Einstein model, all modes have the same frequency ω_E so that

$$g_E(\omega) = n\delta(\omega - \omega_E)$$

where $\delta(x)$ is the Dirac delta function.

• van Hove singularities

Just as for electrons, the density of states can be obtained as an integral over a constant-energy surface S,

$$g(\omega) = \sum_{s} \int \frac{\mathrm{d}S}{(2\pi)^3} \frac{1}{|\nabla \omega_s(\mathbf{K})|}$$

Since $\nabla \omega_s(\mathbf{K})$ may become zero, the slope $dg/d\varepsilon$ displays singularities, also known as *van Hove singularities*.